



# Geologic Sequestration of CO<sub>2</sub>

## Trapping Carbon Dioxide (CO<sub>2</sub>) in the Earth

### Course #101

John Henry Beyer, Ph.D.  
WESTCARB Program Manager, Geophysicist  
510-486-7954, jhbeyer@lbl.gov

Lawrence Berkeley National Laboratory  
Earth Sciences Division, MS 90-1116  
Berkeley, CA 94720



# Course syllabus

- What is CO<sub>2</sub> and how are we making it?
- Why is CO<sub>2</sub> a problem?
- Where would CO<sub>2</sub> go in the earth and how would it be trapped?
- Would there be a big bubble of CO<sub>2</sub>?
- For a small-scale test project, what are we testing?
- What are we not testing and why?
- What's involved in field operations?
- Who else is doing projects like this?



# What and where is carbon dioxide?

- The bubbles in carbonated beverages are CO<sub>2</sub>
- The dry ice in old ice cream trucks was frozen CO<sub>2</sub>
- CO<sub>2</sub> is in the air we breathe



## Sources of CO<sub>2</sub>

- Fire
- Cars, trucks, airplanes
- Electricity generation at coal and gas fired power plants
- Oil refineries
- Cement plants
- Forest fires

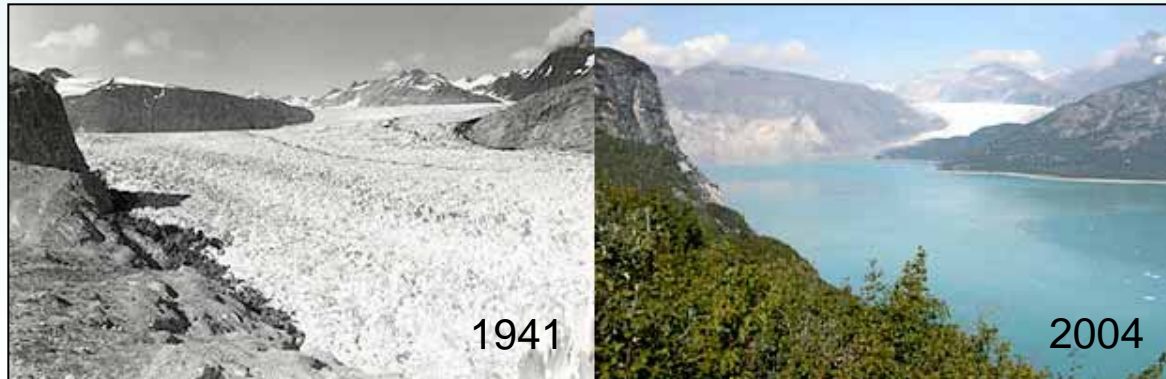


# Why is carbon dioxide a problem?



- CO<sub>2</sub> in the atmosphere acts like the glass in a greenhouse – it traps in heat
- Plants absorb CO<sub>2</sub> and produce oxygen, but we produce more CO<sub>2</sub> than can be absorbed
- The amount of CO<sub>2</sub> in the atmosphere is increasing
- The average temperature of the atmosphere is increasing
- The polar ice caps and glaciers are melting
- Sea level is rising
- Storm frequency and severity is increasing
- Weather patterns are changing
- Many crop lands are threatened

Muir Glacier  
Alaska

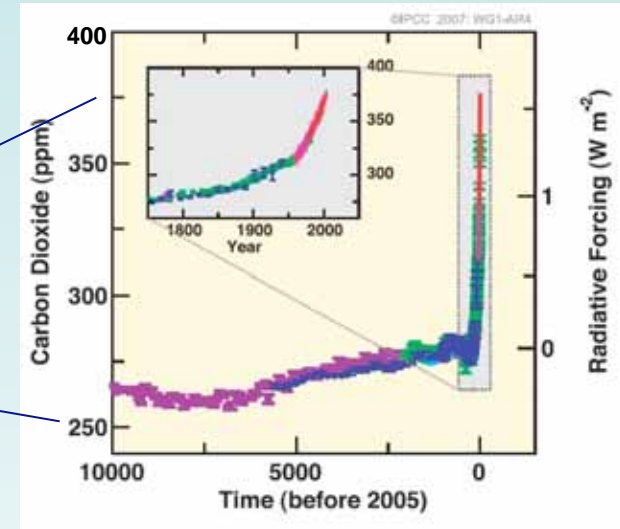
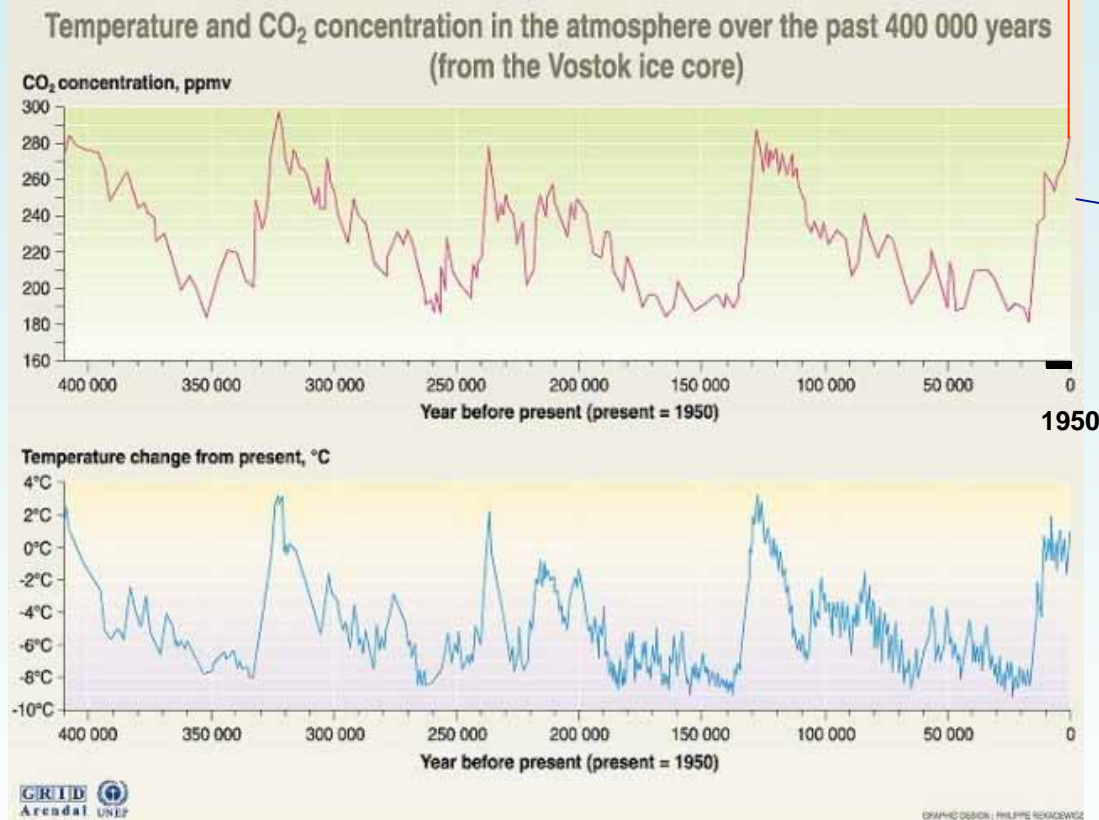


53 years later

Image Credit: National  
Snow and Ice Data Center,  
W. O. Field, B. F. Molnia



# CO<sub>2</sub> concentration and Earth's temperature



Source: IPCC Working Group 1 as presented at <http://www.oneclimate.net/2008/01/15/climate-change-the-truth-greenhouse-gases/>

Source: United Nations Environment Programme. GRID-Arendal.  
Vital Climate Graphics. ISBN: 8277010095.



WEST COAST REGIONAL CARBON SEQUESTRATION PARTNERSHIP

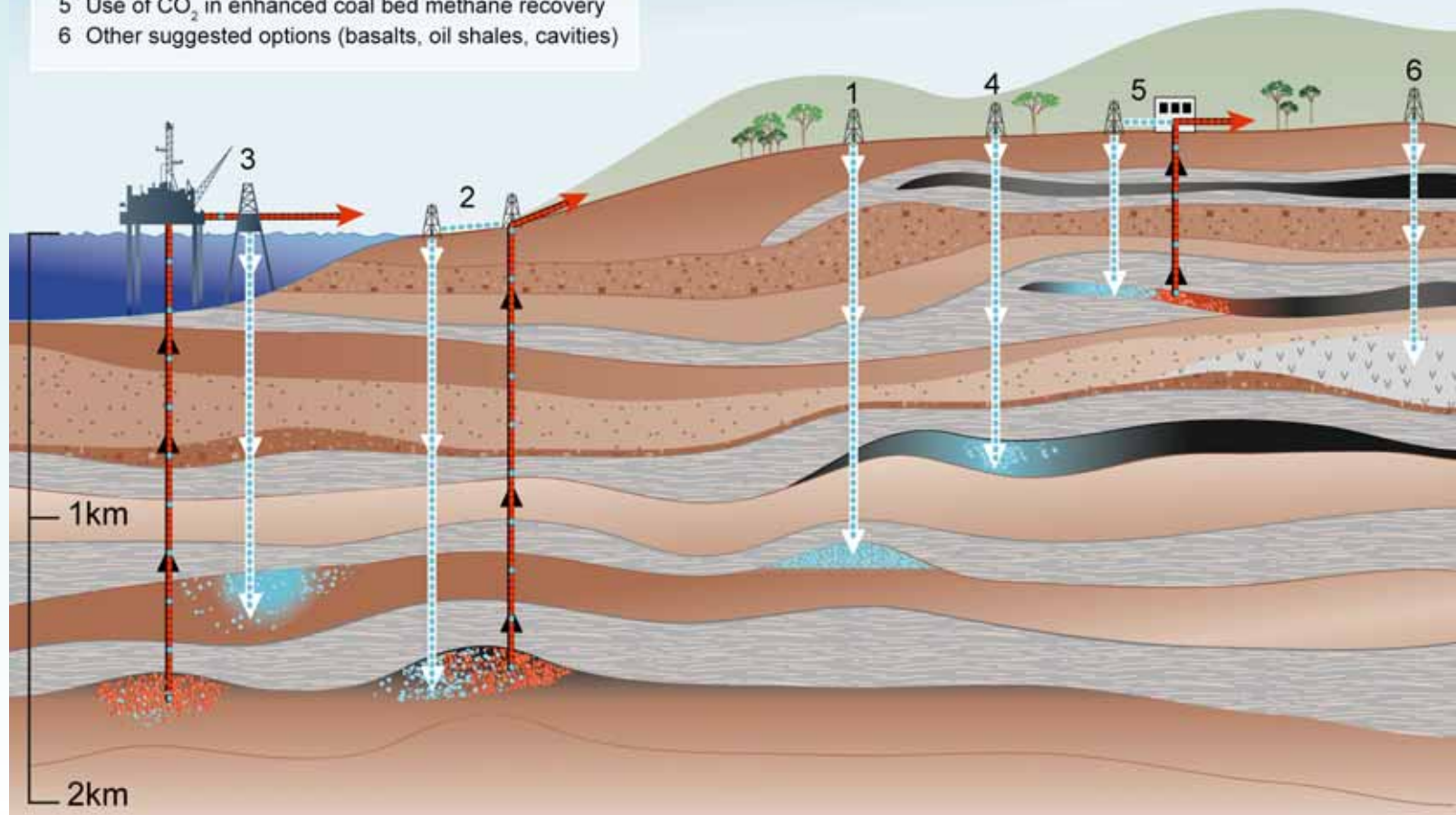


# Geologic CO<sub>2</sub> storage

## Geological Storage Options for CO<sub>2</sub>

- 1 Depleted oil and gas reservoirs
- 2 Use of CO<sub>2</sub> in enhanced oil recovery
- 3 Deep unused saline water-saturated reservoir rocks
- 4 Deep unmineable coal seams
- 5 Use of CO<sub>2</sub> in enhanced coal bed methane recovery
- 6 Other suggested options (basalts, oil shales, cavities)

Produced oil or gas  
Injected CO<sub>2</sub>  
Stored CO<sub>2</sub>



# Important rock types for CO<sub>2</sub> storage

## Cap Rock

Commonly, Shale

- In sedimentary basins it is laterally extensive, thick, and has very fine grains, which makes it relatively impermeable.
- Has held oil and gas in reservoirs for millions of years.



State of Utah, Office of Education,  
[www.usoe.k12.ut.us](http://www.usoe.k12.ut.us)



"Bedford Shale", *Ohio History Central*, July 13, 2007,  
<http://www.ohiohistorycentral.org/entry.php?rec=2857>

## Reservoir (storage) Rock

Commonly, Sandstone

- In sedimentary basins it is laterally extensive, thick, and has high porosity and permeability.
- Most common oil and gas reservoir rocks.



Bureau of Economic Geology,  
University of Texas

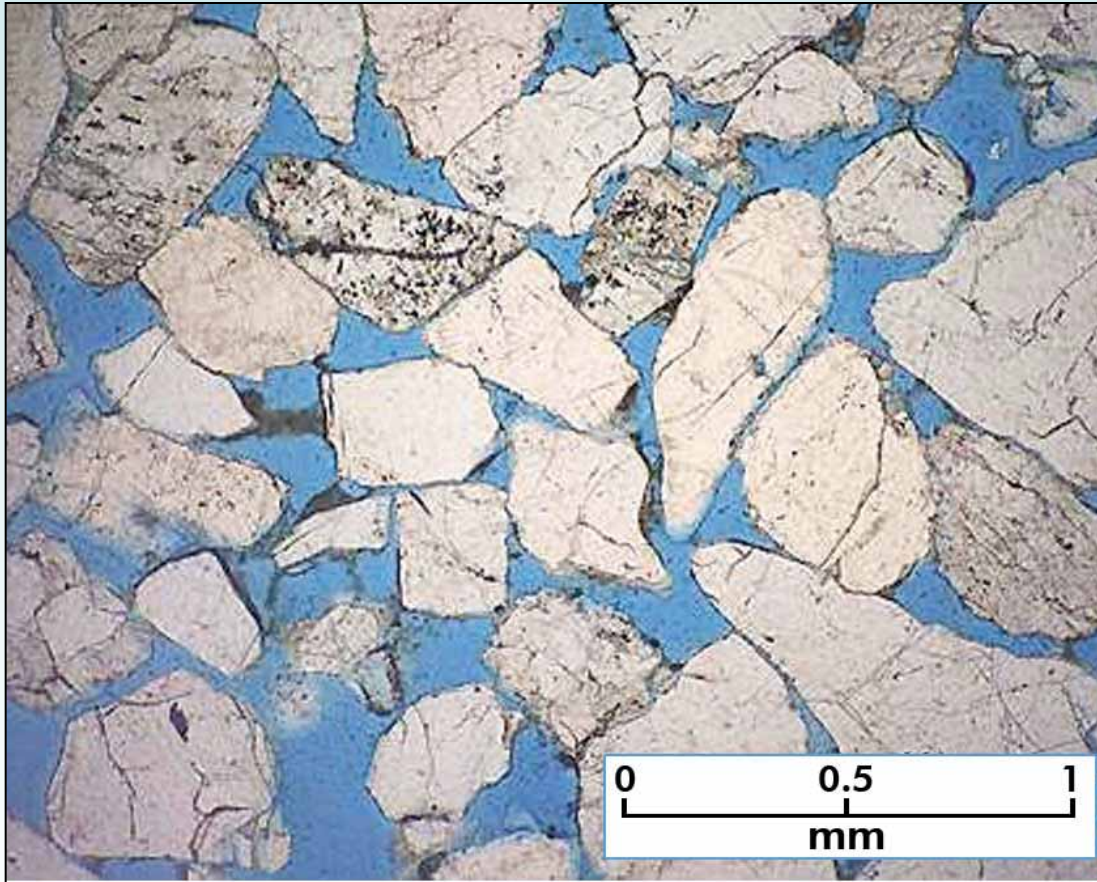


[http://library.thinkquest.org/05aug/00461/  
images/sandstone.jpg](http://library.thinkquest.org/05aug/00461/images/sandstone.jpg)



# Reservoir rock porosity and permeability

Microscopic view of sandstone



The pores in deep rock formations are filled with saline water (brine).

In oil and gas reservoirs, the pore space is filled with hydrocarbons.

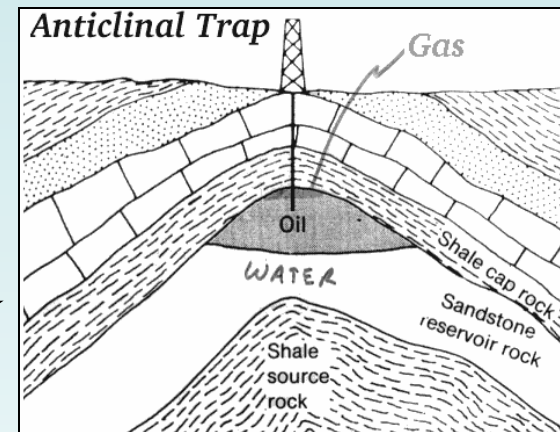
CO<sub>2</sub> both dissolves in and displaces some of the brine.

Source: The Australian Cooperative Research Centre for Greenhouse Gas Technologies (CO2CRC)

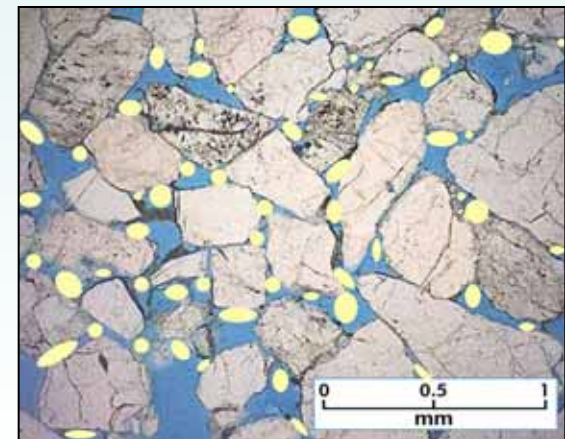


# How CO<sub>2</sub> is trapped deep in the earth

1. Dissolves in saline water in the reservoir rock (like CO<sub>2</sub> dissolved in a soda)
2. Impermeable overlying cap rock and dome-shaped structures (the way oil and gas are trapped)
3. If the plume moves from buoyancy, water fills in behind it, trapping bits of CO<sub>2</sub> in tiny capillaries between the pores of the reservoir rock
4. Chemical combination with minerals dissolved in the formation brine to form new rock

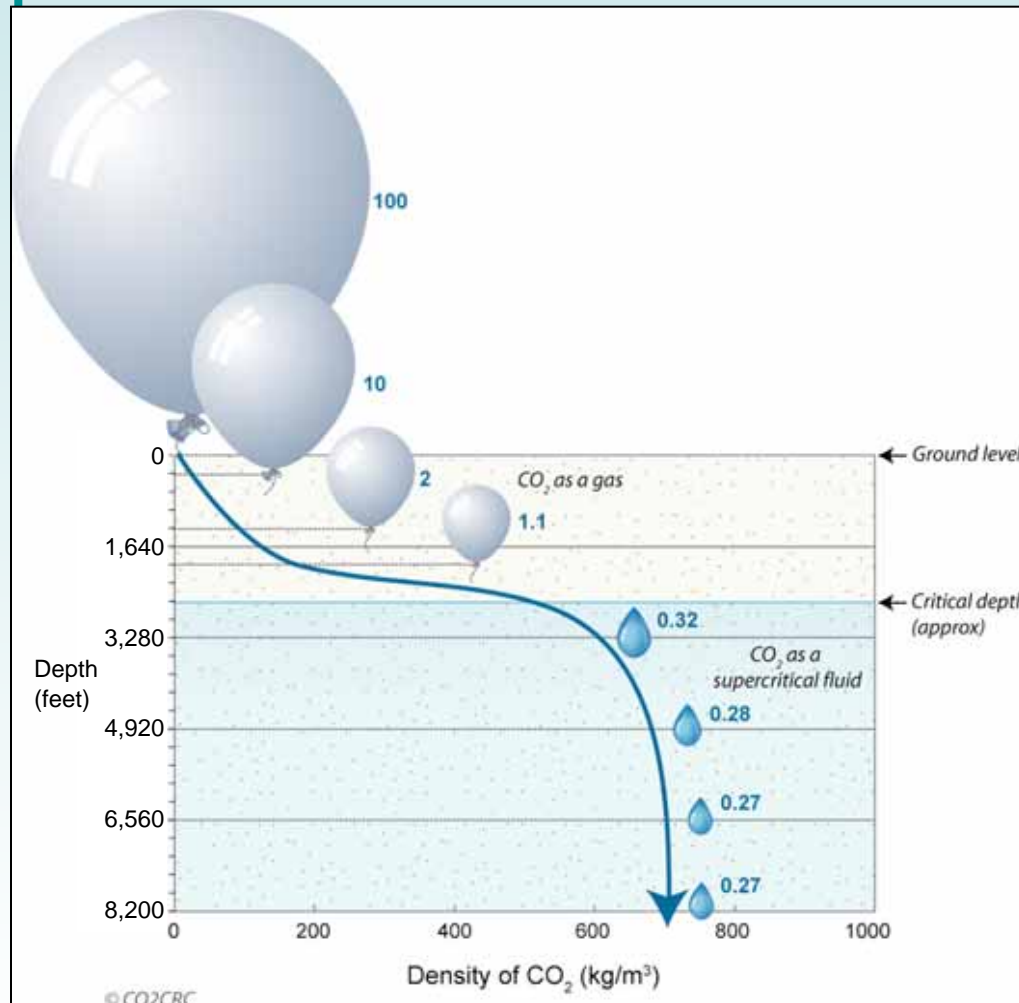


Earth Science Australia, [http://earthsci.org/education/teacher/basicgeol/fossil\\_fuels/anticlinal-trap.gif](http://earthsci.org/education/teacher/basicgeol/fossil_fuels/anticlinal-trap.gif)



# Will there be a big bubble of CO<sub>2</sub> in the ground?

## No - CO<sub>2</sub> is very compressible



Source: The Australian Cooperative Research Centre for Greenhouse Gas Technologies (CO2CRC)

CO<sub>2</sub> compresses by a factor up to 370 from its volume at the surface.

The hydrostatic (water) pressure in the rocks increases by about 1/2 psi per foot of depth in the earth.

So the CO<sub>2</sub> stays compressed by the pressure that naturally exists deep in the earth.

The compressed CO<sub>2</sub> is liquid-like, with about 2/3 the density of water.

# For a small-scale CO<sub>2</sub> injection test project, what are we testing?

- Do the deep potential reservoir rocks have high porosity to contain large quantities of CO<sub>2</sub>?
- Do these reservoir rocks have high permeability so CO<sub>2</sub> can be injected?
- Are there impermeable rock formations or structures above the CO<sub>2</sub> reservoir that will contain the CO<sub>2</sub> indefinitely?
- How will the CO<sub>2</sub> interact with the rock and the saline water in the pores of the rock? - so we can predict when and where the CO<sub>2</sub> plume will stabilize
- What techniques will work best to monitor the movement of the plume until it stabilizes?



## For a small-scale CO<sub>2</sub> injection test project, what are we NOT testing?

- Can we safely drill deep wells and pump large quantities of CO<sub>2</sub> down them?

In the United States today, the oil industry is pumping more than 85,000 tons of CO<sub>2</sub> into the ground every day (31 million tons/year)

- Can we safely transport CO<sub>2</sub> from industrial sources to the wells?

For small quantities, many CO<sub>2</sub> tanker trucks are traveling on our highways every day to supply CO<sub>2</sub> to beverage manufacturers.

For large quantities, the oil industry has a 2,200-mile CO<sub>2</sub> pipeline network that has operated safely for decades.



# Why are the oil companies injecting large quantities of CO<sub>2</sub> into the ground?

## Enhanced Oil Recovery (EOR)

- CO<sub>2</sub> EOR recovers 206,000 BOPD, 12% of lower US oil production
- Started in 1972, it is responsible for more than 1 billion barrels of oil from the Permian Basin
- There are more than 72 US oil fields using the 85,000 tons per day
- The source of CO<sub>2</sub> for EOR is deep geologic formations where nature has stored it for millions of years!



# Field operations for test

## Step 1 – Drill wells

An oil & gas drill rig capable of drilling to several thousand feet is used

This rig is drilling a natural gas well in northwestern California.



Source: Shell

Drill site for Midwest Regional Carbon Sequestration Partnership 9,000-foot well in Edwardsport, Indiana.



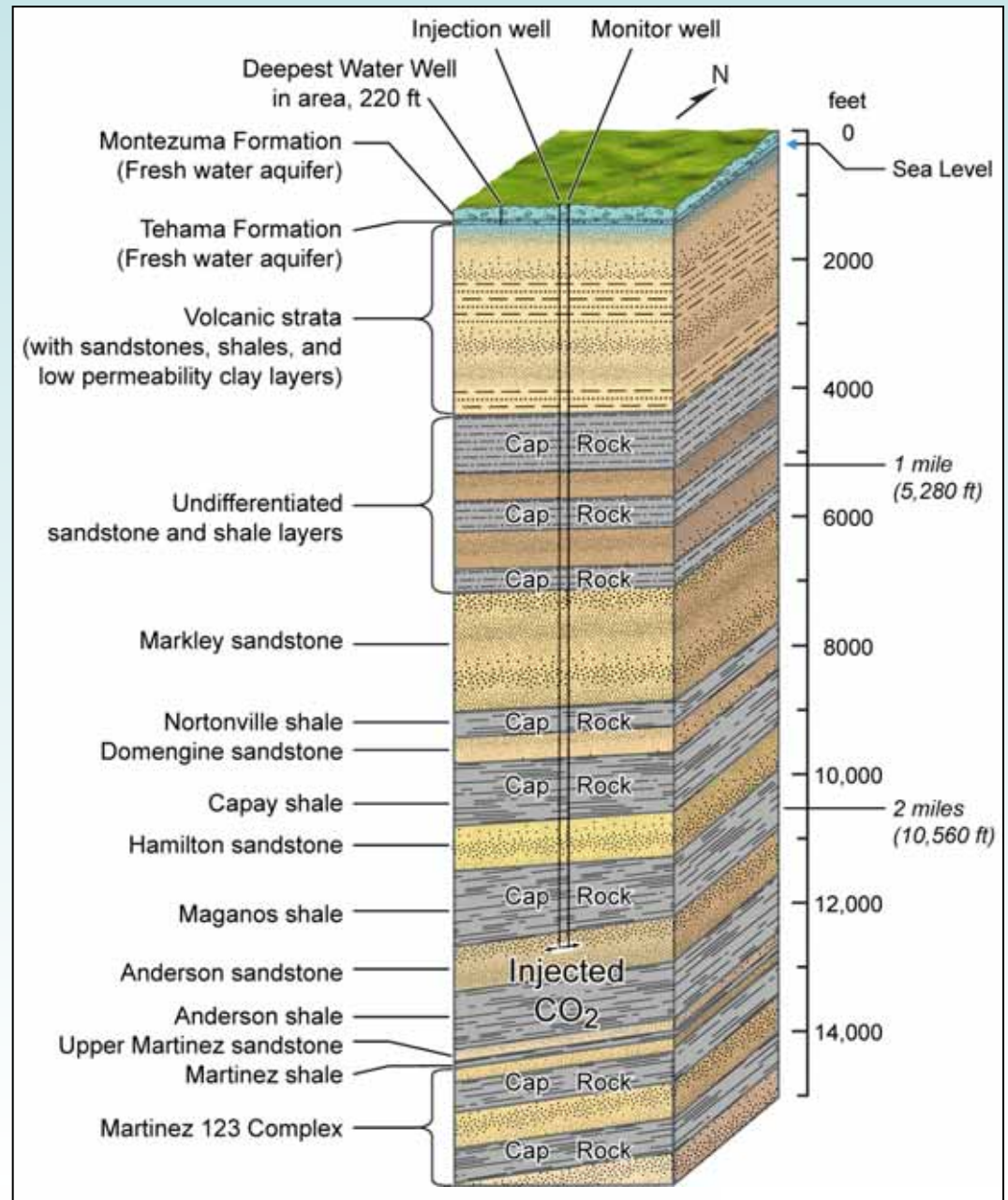
Source: Midwest Regional Carbon Sequestration Partnership,  
David Ball, Battelle



# Field operations

## Step 1 – Drill wells

- Two wells will be drilled more than 2 miles deep and 150 feet apart
- CO<sub>2</sub> will be injected into a permeable sandstone layer beneath multiple impermeable shale layers
- The CO<sub>2</sub> plume is expected to be about 800 feet wide



# Field operations for test

## Step 2 – Inject CO<sub>2</sub>

20-ton tanker trucks will transport CO<sub>2</sub> to the site for injection



The excitement of CO<sub>2</sub> injection



Source: Gulf Coast Carbon Center, part of The University of Texas at Austin's Bureau of Economic Geology

# Field operations for testing

## Stage 3 – Testing and Monitoring

A crane will lower tubing and test equipment into the well



Source: Midwest Regional Carbon Sequestration Partnership

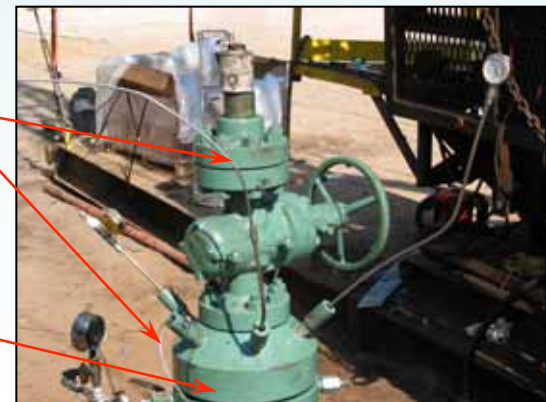
A module with office space, computer, and valves (U-Tube System) will obtain brine samples from the reservoir formation.



U-Tube System at Frio, Texas, CO<sub>2</sub> site

1/4-inch tubes from bottom of well to surface module

Wellhead





## Stage 4 – End of small-scale test project

### Field site after small-scale test project

Wellhead on-site if operator retains the well for future testing or other options



If operator decides to plug & abandon the well, which would require restoring the site



## Stage 5 – Optional - Large-scale injection

Whether this would happen depends on many factors

- Appropriate geologic information from the small-scale test
- Federal laws regulating CO<sub>2</sub> emissions
- Cost of CO<sub>2</sub> emissions
- Nearby sources of CO<sub>2</sub>
- Pipeline from sources to injection site
- Acceptance by regulators and the public



# CO<sub>2</sub> Storage Projects Around the World



Source: The Australian Cooperative Research Centre for Greenhouse Gas Technologies (CO2CRC)